Does sodium bicarbonate increase the amount of vitamin D that gets to cells? Note 1: Apparently, all of the following are known to increase vitamin D in cells: Resveratrol, Omega-3, Magnesium, Zinc, Quercetin, Curcumin, Berberine, Butyrate, Sulforaphane, and Metflormin. Note 2: All of the following genes affect the amount of vitamin D in the blood that gets to the cells. CYP27B1, CYP3A4, CYP24A1; Vitamin D Binding Protein, VDR. Note 3: Only two Vitamin D genes affect how much vitamin D gets measured in the blood: CYP27A1 and CYP2R1

A perplexity

The Impact of Sodium Bicarbonate on Cellular Vitamin D Bioavailability: A Comprehensive Analysis

The relationship between sodium bicarbonate supplementation and vitamin D bioavailability at the cellular level presents a complex and nuanced physiological interaction. Current research offers mixed evidence regarding whether sodium bicarbonate definitively enhances vitamin D delivery to cells, though several studies suggest significant effects on vitamin D metabolism and potential implications for cellular uptake.

Sodium Bicarbonate and Vitamin D Metabolism

Sodium bicarbonate administration appears to influence vitamin D metabolism through multiple pathways, though the resulting effect on cellular vitamin D bioavailability remains incompletely understood. A pivotal animal study demonstrated that bicarbonate (NaHCO3) feeding in rats significantly increased plasma levels of 1,25(OH)2D, the active form of vitamin D, despite simultaneously decreasing duodenal uptake of calcium and phosphate^[1]. This finding suggests that while bicarbonate might reduce intestinal mineral absorption, it potentially enhances the conversion of vitamin D to its active form, which could increase availability to target cells.

Historical animal studies from decades ago established a bidirectional relationship between acidbase balance and vitamin D metabolism. These studies revealed that vitamin D depletion resulted in metabolic acidosis, while vitamin D repletion produced metabolic alkalosis through alterations in renal bicarbonate reabsorption^[2]. Additionally, acute metabolic acidosis was shown to suppress 1- α -hydroxylase activity, thereby reducing the conversion of 25(OH)D to the active 1,25(OH)2D metabolite^[2]. This suggests that alkalization through sodium bicarbonate might theoretically enhance 1- α -hydroxylase activity and increase active vitamin D production.

However, clinical research in humans presents somewhat contradictory results. A single-blind pilot study in patients with chronic kidney disease (CKD) and mild metabolic acidosis found that after six weeks of oral sodium bicarbonate therapy, median FGF23 levels increased significantly from 150.9 RU/mL to 191.4 RU/mL^[3]. Given that FGF23 typically downregulates the production of active vitamin D by inhibiting 1- α -hydroxylase, this finding suggests a potentially negative impact on vitamin D status, contrary to the results observed in animal studies^[3].

Vitamin D Absorption and Cellular Uptake Mechanisms

Understanding vitamin D bioavailability requires consideration of its complex journey from ingestion to cellular action. Vitamin D bioavailability depends on multiple factors including absorption, transportation, and metabolism^[4]. Following intestinal absorption, vitamin D is bound primarily to vitamin D-binding protein (DBP) in plasma, which facilitates its transport to target tissues^[5].

Recent research challenges the traditional view that vitamin D enters cells through passive diffusion. Studies suggest that cellular uptake of vitamin D metabolites occurs predominantly through receptor-mediated endocytosis of vitamin D bound to DBP or albumin^[6]. This uptake mechanism involves plasma membrane receptors like megalin (LDL-related protein 2) and cubilin, which mediate vitamin D internalization through clathrin-dependent endocytosis^[6]. These findings suggest that factors affecting these uptake mechanisms could potentially influence cellular vitamin D bioavailability.

The bioaccessibility of vitamin D—its availability for absorption in the gastrointestinal tract—varies significantly between different forms and food matrices. Vitamin D3 appears to be more bioaccessible from food sources than from supplements, with its bioaccessibility particularly susceptible to changes in gastric pH^{[7] [8]}. Importantly, exposing vitamin D to lower gastric pH during digestion significantly enhances its bioaccessibility in the intestinal phase^[8]. This raises interesting questions about how sodium bicarbonate, which increases pH, might affect vitamin D absorption.

The Potential Impact of Sodium Bicarbonate on Cellular Vitamin D

While direct evidence for sodium bicarbonate increasing cellular vitamin D uptake remains limited, several potential mechanisms warrant consideration. Sodium bicarbonate's alkalinizing effect could theoretically enhance 1- α -hydroxylase activity, potentially increasing the production of active 1,25(OH)2D^[2]. However, this must be balanced against the evidence that bicarbonate supplementation increases FGF23^[3], which typically inhibits 1- α -hydroxylase.

Sodium bicarbonate's effect on gastric pH is another important consideration. Research demonstrates that vitamin D3 bioaccessibility is enhanced in more acidic conditions, with higher intestinal content of vitamin D3 observed when samples are exposed to lower gastric pH^[8]. Since sodium bicarbonate raises gastric pH, it might theoretically reduce vitamin D absorption, counteracting any potential benefit from enhanced activation.

The impact of sodium bicarbonate on vitamin D binding proteins and cellular uptake mechanisms represents another area where effects might occur. The binding of vitamin D to DBP affects its bioavailability, with different DBP variants showing varying binding affinities for vitamin D metabolites^[5]. If sodium bicarbonate affects the binding affinity or concentration of these transport proteins, it could potentially influence cellular vitamin D delivery.

Genetic Factors Influencing Vitamin D Metabolism and Cellular Uptake

As noted in the query, several genes play crucial roles in vitamin D metabolism and cellular bioavailability. CYP27B1 encodes the 1- α -hydroxylase enzyme responsible for converting 25(OH)D to active 1,25(OH)2D in the kidneys and other tissues^{[5] [6]}. CYP24A1 encodes 24-hydroxylase, which metabolizes vitamin D for excretion, and CYP3A4 is involved in vitamin D catabolism^[9]. Additionally, variations in the vitamin D binding protein gene can affect transport and bioavailability^[5]. The vitamin D receptor (VDR) is essential for cellular responses to vitamin D, mediating both genomic and non-genomic effects^{[5] [10]}.

The presence of CYP27B1 and VDR in multiple tissues beyond the kidneys suggests that local vitamin D activation and signaling occur in various cells throughout the body^[10]. This localized vitamin D system means that factors affecting vitamin D metabolism could have tissue-specific effects that might not be reflected in circulating vitamin D levels.

Conclusion

The evidence regarding sodium bicarbonate's effect on cellular vitamin D bioavailability presents a complex picture. While animal studies suggest that bicarbonate supplementation increases plasma levels of active vitamin D^[1], human studies indicate that it also increases FGF23^[3], which typically reduces active vitamin D production. Furthermore, sodium bicarbonate's alkalinizing effect might enhance vitamin D activation through 1- α -hydroxylase^[2] but could potentially reduce absorption by increasing gastric pH^[8].

Based on the available evidence, sodium bicarbonate appears to influence vitamin D metabolism through multiple pathways, but whether these effects ultimately increase the amount of vitamin D that reaches cells remains uncertain. The interaction likely depends on numerous factors including baseline vitamin D status, renal function, genetic variations in vitamin D metabolism enzymes and transport proteins, and the specific physiological context. Further research specifically examining cellular vitamin D content following sodium bicarbonate administration would be necessary to definitively answer this question.

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